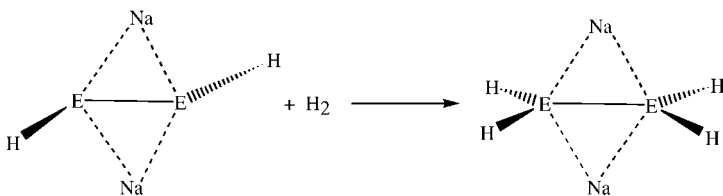


Reaction enthalpies are calculated for the hydrogenation reactions of main group hydrides with the potential for multiple bonding, and thus the *unsaturated* character of these species was determined. The study includes species such as $[\text{HGaGaH}]^{2-}$ and $\text{Na}_2[\text{HGaGaH}]$ (see scheme, E = B, Al, or Ga).



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Heats of Hydrogenation of Compounds Featuring Main Group Elements and with the Potential for Multiply Bonding



Supporting information on the WWW (see article for access details).

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CORRIGENDA

In the paper by K. C. Nicolaou et al. published in *Chem. Eur. J.* **2001**, 7, 5359–5371, the enantiomeric structures of the Diels–Alder components in the transition states TS_a and TS_b are required to account for the observed enantiomers **8**, (–)-**5**, **8'**, and **5'** shown in Scheme 9.

In the paper by J. Jiang, N. Kobayashi, and D. K. P. Ng et al. in *Chem. Eur. J.* **2001**, 7, 5059–5069, the true space group for all the double-deckers **4**·0.5 C_6H_{12} , **6**, **9**·0.5 C_6H_{12} , and **14** is *Pnam* (No. 62; interchange of the *b* and *c* axes as reported in the paper leads to the standard symbol *Pnma*). The $[\text{M}^{\text{III}}(\text{nc})(\text{oep})]$ molecule lies on a crystallographic mirror plane that contains the lanthanide atom, two *trans*-related isoindole N donor atoms of the nc ring, and a pair of *trans*-related *meso* CH groups on the oep ligand. In the complexes **4**·0.5 C_6H_{12} and **9**·0.5 C_6H_{12} , the chair-shaped cyclohexane molecule has half-site occupancy with two of CH_2 groups lying in a mirror plane. The authors apologize for these errors.